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the nucleus remember that the nucleus is comprised of the two nucleons protons and neutrons the number of protons is the atomic number the number of protons and neutrons together is effectively the mass of the atom isotopes not all atoms of the same element have the same mass due to different numbers of neutrons in those atoms

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21 1 1 nuclear equations alpha particles helium 4 particles alpha radiation stream of alpha particles emission of radiation is one way that an unstable nucleus is transformed into a more stable one superscript mass number subscript atomic number radioactive decay when a nucleus spontaneously decomposes

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nuclear chemistry is the study of nuclear reactions their uses in chemistry and their impact on biological systems when nuclei change spontaneously emitting energy they are said to be radioactive

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a nuclear reactor consists of the following 1 a nuclear fuel a fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction the radioactive isotope is contained in tubes called fuel rods 2 a moderator

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an international atomic energy agency iaea team of experts completed a review of long term operational safety at unit 1 of the angra nuclear power plant in brazil the safety aspects galateo bon ton moderno ma non troppo le buone maniere ieri e oggi
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of long term operation salto review mission was requested by the eletronuclear s a the public entity that owns and operates angra npp

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the difference between the mass of an atom and the sum of the masses of its protons neutrons and electrons nuclear binding energy the energy released when a nucleus is formed from nucleons a measure of the stability of a nucleus binding energy per nucleon

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