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when radical polymerization is desired it must be started by using a radical initiator such as a peroxide or certain azo compounds the formulas of some common initiators and equations showing the formation of radical species from these initiators are presented below

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one mechanism to achieve polymerization is radical polymerization a chain reaction that is started with the introduction of an initiator with a free radical the initiator is shown below as r more important than its chemistry is the free radical a highly reactive single electron

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this chapter introduces the fundamental concepts of radical polymerization typical monomers initiators and transfer and termination reagents are discussed furthermore the kinetic equations the degree of polymerization and molar mass

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summary this chapter deals with the description of the free radical polymerization frp mechanism including initiation propagation and bimolecular termination reaction steps additional reactions such as chain transfer are introduced to improve the process description

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the mechanochemically controlled radical polymerization mechano crp not only facilitate the synthesis of polymers with high molecular weight but also enable precise control over polymer chain length and structure

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strictly speaking this chapter is on radical reactions which include most addition polymerization processes however in order to cover the topic of polymers properly we are including a brief section on condensation polymerization usually based on substitution chemistry here as well

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since its inception atom transfer radical polymerization atrp has seen continuous evolution in terms of the design of the catalyst and reaction conditions today it is one of the most useful techniques to prepare well defined polymers as well as one of the most notable examples of catalysis in polymer chemistry

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