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gas molecules move randomly in their containers colliding with the walls of their containers causing gas pressure pressure can be defined as the force pushing on an area it can be described with the equation $p = \frac{F}{A}$ where p is pressure in kpa F is force in n and A is area in m^2

1 write the formulas for the following compounds
 It n dinitrogen pentoxide el i jr calcium phosphide copper il phosphate ut utt u ele d lithium sulfate i 2 write the names for the foliowing compounds a ca cn 2 c feco3 o rru c r n n c6 s son l lg r rr ksu lu b pzo

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skill practice 35 name date hour important whenever you use temperature it must be in degree kelvin k so remember the equation $k = ^\circ C + 273.15$ a convert $39^\circ C$ to k b convert 127 k to $^\circ C$ 312 k $146^\circ C$ 2 a gas has an initial volume of 2.75 l at a temperature of 285 k

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answer the questions that follow based on this chemical equation $\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$ a why can't NaMg be produced hint consider the charges of the sodium and magnesium ions b why can't NaCl₂ be produced c given your answers to parts a and b do you think that NaCl and Mg the only products that can be produced

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in the introductory questions hopefully you have discovered that you need two pieces of information to know how concentrated a solution is 1 the amount of solute dissolved and 2 the volume of the liquid here s a formula to calculate the concentration of a solution

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chemquest 37 sslgdcll g e a1ue date hour i ntroduction questions information subscripts and coefficients a subscript is a small number that tells you how many atoms are in a compound for example in CaCl_2 the two is the subscript and it tells us that there are two chloride ions bonded to one calcium

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