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some kind of liquid solutions in normal life we  
rarely come across pure substances most of these  
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solution a solution is a homogeneous mixture comprising smaller components called solutes of small molecules or ions comparable in size to the molecules of a larger component called the solvent for example NaCl dissolved in water is a solution

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explain why solutions of HBr in benzene a nonpolar solvent are nonconductive while solutions in water a polar solvent are conductive answer HBr is an acid and so its molecules react with water molecules to form  $\text{H}_3\text{O}^+$  and  $\text{Br}^-$  ions that provide conductivity to the solution

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a solution is a homogeneous mixture of two or more substances and can either be in the gas phase the liquid phase the solid phase the enthalpy change of solution refers to the amount of heat that is released or absorbed during the dissolving process at constant pressure

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