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reactions unit1 acids and bases unit2 why metals are awesome unit 3 why carbon is responsible for life unit 4 and how we classify elements unit 5 unit 1 chemical reactions and equations 0 1200 mastery points applied chemistry close to our daily life we are aiming to design molecules and materials in various ways to find their new functions our wide range of various cutting edge researches goes beyond the ordinary categorization of sciences such as inorganic or organic chemistry and biotechnology

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use the prefix that applies to your situation this table of metric prefixes has the standard names for each prefix used in science and measurements it also shows the symbol used to represent these prefixes their power of ten and the number you are actually replacing with the prefix table of powers of 10 prefixes

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learning objectives explain the process of measurement and describe the three basic parts of a quantity describe the properties and units of length mass volume density temperature and time recognize the common unit prefixes and use them to describe the magnitude of a measurement

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a type of reaction that occurs when hydrocarbon reacts with oxygen gas to produce carbon dioxide and water it is also known as burning if there is enough oxygen hydrocarbon burn completely to release the energy they contain the only products of complete combustion are carbon dioxide and water

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