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solutions in which more solute is dissolved than can normally occur at a given temperature a gas s solubility in a liquid is directly proportional to its pressure

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step 1 of 1 the vapor pressure lowering is governed by raoult s law it states that the vapor pressure of the solvent above the solution is equal to vapor pressure of the pure solvent at that temperature multiplied by the mole fraction of the solvent present the expression is given as follows

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


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