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1 covalent bond a covalent bond in which electrons are not shared equally 3 valence atoms gain more than 8 valence electrons 4 numbers shows how many electrons can be shared lost or gained to become stable you cancel out the valence numbers to make the compound 5 polarity the uneven distribution of molecular charge

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an element that has gained or lost an electron a negatively charged ion a positively charged ion the energy required to pull apart the atoms in a chemical bond the distance between nuclei of 2 bonded atoms show

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we were part of the inaugural pairs competition at the 46th annual tournament a festive three day event in stamford connecticut with almost a thousand competitors jamming the marriott for puzzle themed events leading up to the finals on sunday one skinny contestant came dressed as a pencil a popular young woman wore a crosshatched crossword

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