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## covalent bond wikipedia

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## covalent bond wikipedia

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a covalent bond is a chemical bond that involves the sharing of electrons to form electron pairs between atoms these electron pairs are known as shared pairs or bonding pairs the stable balance of attractive and repulsive forces between atoms when they share electrons is known as covalent bonding 1

## covalent bond definition properties examples facts

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covalent bond in chemistry the interatomic linkage that results from the sharing of an electron pair between two atoms the binding

arises from the electrostatic attraction of their nuclei for the same electrons a covalent bond forms when the bonded atoms have a lower total energy than that of widely separated

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□□□ the best guide to the covalent or ionic character of a bond is to consider the types of atoms involved and their relative positions in the periodic table bonds between two nonmetals are generally covalent bonding between a metal

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□□□ this is called a covalent bond electronegativity is a measure of how strongly an atom attracts electrons from another atom in a chemical bond and this value is governed by where the particular atom is located in the periodic table francium is the least electronegative element while fluorine is the most electronegative example of a

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□□□ 2020□7□30□ the sharing of electrons between atoms is called a covalent

bond and the two electrons that join atoms in a covalent bond are called a bonding pair of electrons a discrete group of atoms connected by covalent bonds is called a

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2023-08-26 covalent bonds form when electrons are shared between atoms and are attracted by the nuclei of both atoms in pure covalent bonds the electrons are shared equally in polar covalent bonds the

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2019-07-03 a covalent bond in chemistry is a chemical link between two atoms or ions in which the electron pairs are shared between them a covalent bond may also be termed a molecular bond covalent bonds form between two nonmetal atoms with identical or relatively close electronegativity values

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