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these lecture notes are intended to supplement a graduate level course in condensed matter physics

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in fact condensed matter physics is by far the largest single subfield of physics the annual meeting of condensed matter physicists in the united states attracts over 6000 physicists each year

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bonding is associated with a spatial function that is symmetric under exchange of the two electrons in this state the electron density is largest between the atoms the total function must be antisymmetric since electrons are fermions so the spin function must be antisymmetric

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in a physics lab you measure the vibrational rotational spectrum of hcl the estimated separation between absorption peaks is $\Delta f = 5.5 \times 10^{11}$ hz the central frequency of the band is $f = 0.9 \times 10^{13}$ hz

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matter material substance that constitutes the observable universe and together with energy forms the basis of all objective phenomena at the most fundamental level matter is composed of elementary particles known as quarks and leptons the class of elementary particles that includes electrons

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matter is anything that occupies space and has mass the three states of matter are solid liquid and gas a physical change involves the conversion of a substance from one state of matter to another without changing its chemical composition

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soft matter consists of units much larger than atoms their typical length is $0.01\text{ }\mu\text{m}$ $100\text{ }\mu\text{m}$ this gives two characteristics to soft matter large non linear response to external forces and long relaxation time and slow dynamics

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matter is the substance which occupies space and has mass the three states of matter are solid liquid gas matter is composed of large number of molecules a molecule is the smallest particle which can exist freely in nature by itself and it retains the properties of the substance

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