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4 7 explain the difference between a strong acid and a weak acid acids are molecular compounds that ionize when they dissolve in water a strong acid completely ionizes in solution a weak acid does not completely ionize in water a solution of a weak acid is composed mostly of nonionized acid

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a n is a full s and p sublevel eight electrons that competes the outer shell of an atom and makes the element very stable and chemically nonreactive octet is the measure of the size of an ion a periodic trend of elements ionic radius

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if you put pure magnesium in your hand it may react with the water in your skin and ignite but it won t melt if you could win a million dollars answering 20 chemistry questions would you be able to win the prize here s your chance to find out

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this online quiz is intended to give you extra practice in naming chemical compounds including ionic compounds made from simple or polyatomic ions binary molecular compounds or common acids select your preferences below and click start to give it a try

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question what is the branch of chemistry which deals with the study of chemical reaction rate answer chemical kinetics is the branch of physical chemistry that is concerned with understanding the rates of chemical reactions it is to be contrasted with thermodynamics which deals with the direction in which a process occurs but in itself

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quiz for students covering some chemistry topics taught in key stage 3 questions and answers 1 what is the ph of a weak acid a 1 b 5 c 7 d 12 correct answer b 5 explanation a weak acid has a ph value less than 7

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chapter 3 form 4 1 multiple choice a balloon contains  $6.02 \times 10^{23}$  of gas particles what is the number of moles of the gas in the ballon 2 multiple choice the mass of eight atoms of element x is equal to the mass of two silicon atoms x is not the actual symbol of the element

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