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balance the equation $2Al + 3Cu(NO_3)_2 \rightarrow 2Al(NO_3)_3 + 3Cu$ a subscript in a polyatomic ion moves with the ion so the 3 in NO_3 stays with the ion but the subscript 2 is there only to balance the charges it's not part of the ion and doesn't move with it write the balanced equation for the following reaction

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