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mixtures can be classified as heterogenous mixtures or as homogenous mixtures how can mixtures be separated differences in physical properties can be used to separate mixtures

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a mixture in which the components are so finely mixed that the composition is the same throughout

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mixtures class section review objectives classify a sample of matter as a substance or a mixture distinguish between homogeneous and heterogeneous samples of matter describe two ways that components of mixtures can be separated vocabulary

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describe the difference between homogenous and heterogenous mixtures distinguish a homogenous mixture as a solution or colloid as introduced previously mixtures are combinations of two or more substances that each retain their individual physical properties

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how can mixtures be classified a salad bar like the one in figure 2 5 provides a range of items such as let tuce tomatoes cheese and green peppers customers choose which items to use in their salads and how much of each item to use so each salad mix how can mixtures separated

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lesson 1 types of matter objective differentiate between compounds mixtures and elements determine if a mixture is homogeneous or heterogeneous classify each of the following with the combination of terms listed below pure substance element mixture homogeneous pure substance compound mixture heterogeneous hcl aq

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consider a mixture of two volatile liquids a and b the vapor pressure of pure liquid a is 324 3 torr and that of pure liquid b is 502 3 torr what is the total vapor pressure over a mixture of the two liquids for which x b 0 675

types of mixtures homogeneous and heterogeneous mixtures *May 07* 2023

google classroom microsoft teams which among the following statements is true about pure substances choose 1 answer these are mixtures of different elements a these are mixtures of different elements pure substances are those in which all the molecules are of the same nature b

solved 22 mixtures of solutions calculate the ph of the Apr 06 2023

question 22 mixtures of solutions calculate the ph of the solutions made as indicated assume the volumes are additive a 60 ml of 0 12 m hcn 40 ml of 0 20 m naoh b 50 ml of 0 06 m nh3 60 ml of 0 05 m hci a hcn naolt 2 na n h2o b nh3 hcl 2 nhu nily c wb ab hy 1 8 105 date i need the answer without using the log method

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purpose a heterogeneous mixture of iron fe salt nacl and sand sio2 will be separated by physical means and the percent composition of each will be determined background matter can be classified by composition as a pure element pure compound homogeneous mixture or heterogeneous mixture

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problem set 22 mixtures of gases a what is meant by standard temperature and pressure stp temperature of 273 k 0 degrees c 32 f and an absolute pressure of exactly 100 00 pa 1 bar 14 psi 0 atm b what is the volume of 1 mole of an ideal gas at stp does it matter what gas

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